

Syllabus: Mater Structure.

**UNIVERSITY OF PUERTO RICO  
RÍO PIEDRAS CAMPUS  
COLLEGE OF NATURAL SCIENCES  
DEPARTMENT OF PHYSICS**

Title: Mater Structure

Code: FISI 8105

Number of Credits: 3

Prerequisites: Permission of the Graduate Committee

**August 2016**

**Description**

Mater Structure. Development of the basic formalism including the various representations and 'pictures'. Basis of Mater Structure. Basis approaches developed in Mater Structure. Application of tools and methods to analyze structure of simple and complex atomic and molecular systems, and dynamics and structure of biophysical/chemical processes.

**Objectives**

Through this course, the students will:

- Acquire a basic understanding of the laws of Mater Structure.
- Become proficient with the mathematical formalism of Mater Structure analysis.
- Develop problem-solving skills and strategies in basic Mater Structure in Physics, Chemistry and Biology.
- Communicate effectively a topic pertinent to basic Mater Structure Statements in Physics, Chemistry and Biology

**Course Contents and Time Distribution**

- I. Introduction, Quantum Mechanics Basis (2 wks)
  - a) Schrodinger equation;
  - b) The particle in the box;
  - c) Stationary and not stationary states;
  - d) The perturbation stationary and not stationary theory;
  - e) The variation method;
  - f) Spin;
  - g) The Pauli principle;
  - h) Symmetry point groups;
  
- II. Electronic Structure and Spectroscopy of Atoms (3 wks)
  - a) Hydrogen atom as model system in atomic spectroscopy.
  - c) Multielectronic atoms;
  - d) Single electronic approximation;
  - e) Atomic terms. Russell-Saunders and  $j-j$  angular momenta coupling schemes;
  - f) Fine and hyperfine state structures;

- g) Stark effect in atomic states;
- h) Zeeman effect in atomic states..

III. Diatomic molecules (2 wks)

- a) Systematics of the diatomic molecules terms. .
- b) Correlation between atomic terms of the separated atoms with diatomic molecule terms;
- c) Borhn-Oppenheimer approximation;
- d) Hitler-London method: method of valence scheme (VS);
- e) Selection rules for dipole approximation;
- f) Molecular Orbital method (MO);
- g)  $H_2^+$  ion state structure;
- h) Chemical bond. Komton's and exchange integrals;
- i) Different bond nature:  $\sigma$ - and  $\pi$ -bonds;
- j) Comparison of VS and MO methods.

IV. Electronic Structure and Electronic Spectra of Polyatomic Systems (2 wks)

- a) Group Theory;
- i) MO method. Hartry-Fock approximation;
- j) MO of  $\pi$ -bond systems;
- k) MO Huckel method (MOH);
- l) MOH for systems with heterogeneous atom;
- m) Charge distribution in  $\pi$ -bond systems;
- n) Spin density distribution in  $\pi$ -bond systems.

V. Reaction Activity of Molecular systems (1 wks)

- a) Perturbation method application to MO;
- b) Reactions between  $\pi$ -radicals;
- c) Woodward-Hoffman rule. Conservation of orbital symmetry;

VI. Electronic structure and Spectroscopy of Coordinative Inorganic Systems (1 wk)

- a) Crystal field theory;
- b) MO theory application to coordinative inorganic systems;
- c) State structure of coordinative inorganic systems;
- d) Spectroscopy of coordinative inorganic systems.

VI. Application of other spectroscopic methods to study of mater structure (4 wks)

- a) State structure in low energy region;
- b) Radiofrequency and Microwave spectroscopy;
- c) State structure in intermediate energy region ( $k_B T < E < 1 \text{ eV}$ );
- d) Vibration spectroscopy.

**Instructional Strategies**

Lectures, problem sets, group discussions, and oral presentations by the students.

**Minimum Required Facilities**

Traditional lecture room

### **Student Evaluation**

Two partial exams, 3 homework assignments, and oral presentation

### **Grading System**

Standard A to F grading system:

100-90% = A, 89-80% = B, 79-70% = C, 69-60% = D, 59-0% = F.

### **Bibliography**

1. J.N. Murrell, S.F.A. Kittle, J.M. Tedder, Valence Theory, John Wiley and Sons PLD, London – New York - Sydney – Toronto, 1969.
2. L. D. Landau and E. M. Lifshitz, Quantum Mechanics, Fizmatgis, Moscow (1963).
2. William A. Guillory, Introduction to Molecular Structure and Spectroscopy., **Allyn and Bacon; 1st edition (1977)**

### **Online Resources:**

- [https://www.unf.edu/~michael.lufaso/chem4627/ch1\\_solid\\_state.pdf](https://www.unf.edu/~michael.lufaso/chem4627/ch1_solid_state.pdf)
- <http://scuseria.rice.edu/gau/PBC-Guide.pdf>
- <https://kar.kent.ac.uk/2546/1/Host-guest.pdf>

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